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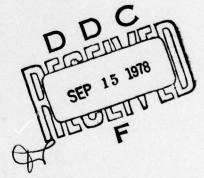
SUMMARY OF UPPER ATMOSPHERIC DATA

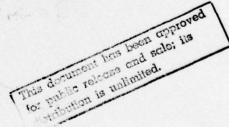




N. Sundararaman

October 1976





Prepared for

U.S. DEPARTMENT OF TRANSPORTATION
Federal Aviation Administration
Office of Environmental Quality
Washington, D.C. 20591

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Operations Research, Inc. 1400 Spring Street	
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2. Sponsering Agency Name and Address High Altitude Pollution Program	(9) Data Repert, 5
Office of Environmental Quality Federal Aviation Administration	
Washington, D.C. 20591	
5. Supplementary Notes	
Simultaneously observed concentrations of NO ₂ , HNO ₃ , and H ₂ O are reported. Also inc CF2C13, CFC12 and N ₂ O, and photoabsorption ClONO ₂ , and NC ₂ .	Cl and ClO; NO, NO ₂ , and HNO ₃ ; and luded are mixing ratios of HCl, cross sections for CCl ₂ F ₂ , CCl ₃ F,
Simultaneously observed concentrations of	Cl and ClO: NO. NO. and HNO.; and
Simultaneously observed concentrations of NO2, HNO3, and H20 are reported. Also inc CF2C13, CFC12 and N20, and photoabsorption	Cl and ClO: NO. NO. and HNO.; and
Simultaneously observed concentrations of NO2, HNO3, and H20 are reported. Also inc CF2C13, CFC12 and N20, and photoabsorption	Cl and ClO: NO. NO. and HNO.; and
Simultaneously observed concentrations of NO2, HNO3, and H20 are reported. Also inc CF2C13, CFC12 and N20, and photoabsorption	Cl and ClO: NO. NO. and HNO.; and
Simultaneously observed concentrations of NO2, HNO3, and H20 are reported. Also inc CF2C13, CFC12 and N20, and photoabsorption	Cl and ClO: NO. NO. and HNO.; and
Simultaneously observed concentrations of NO2, HNO3, and H20 are reported. Also inc CF2C13, CFC12 and N20, and photoabsorption	Cl and ClO: NO. NO. and HNO.; and
Simultaneously observed concentrations of NO ₂ , HNO ₃ , and H ₂ O are reported. Also inc CF2Cl ₃ , CFCl ₂ and N ₂ O, and photoabsorption ClONO ₂ , and NC ₂ .	Cl and ClO: NO. NO. and HNO.; and
Simultaneously observed concentrations of NO ₂ , HNO ₃ , and H ₂ O are reported. Also inc CF ₂ Cl ₃ , CFCl ₂ and N ₂ O, and photoabsorption ClONO ₂ , and NC ₂ . 17. Key Words Absorption cross sections Mixing ratios Atmosphere Solar fluxes Stratosphere Data Trace gases	Cl and ClO; NO, NO2, and HNO3; and Luded are mixing ratios of HCl, cross sections for CCl2F2, CCl3F,

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Introduction

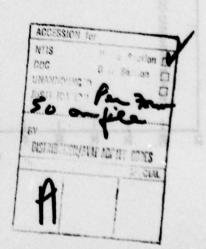
The following data related to the upper atmosphere are presented in this report:

- 1. Concentrations of O(3P), OH
- 2. Concentrations of Cl and ClO observed simultaneously
- 3. Volume and mass mixing ratios of HC1
- 4. Volume mixing ratios of CF2Cl3, CFCl2 and N2O
- 5. Concentrations, simultaneously observed, of NO, NO2 and HNO3; and of NO2, $\rm HNO_3$ and $\rm H_2O$
- 6. Photoabsorption cross sections, and their temperature dependences, for CCl_2F_2 and CCl_3F
- 7. Photoabsorption cross sections of ClONO2 and of NO2.

The data are displayed in graphical form, and, if readily available, in tabular form. Some comments have been included regarding the observational technique, date, time and location of the experiment and the associated uncertainty. Lists of references are appended immediately after these comments or after the data.

In the "Initial Summary of Upper Atmospheric Data," disseminated in April 1976, two errors have been noted: one, in the improper labeling of the abscissae in the HCl absorption cross section diagram, and the other in the wrong copying of the table of values for the 5 Å - average solar flux values for quiet sun conditions in the 1750-2100 Å region. Both of these oversights have been corrected and the corrected diagram (Figure 17) and table (Table 10) are included in this report.

It is quite likely that the author of this report has missed some of the data; he would be grateful if such data are brought to his attention.



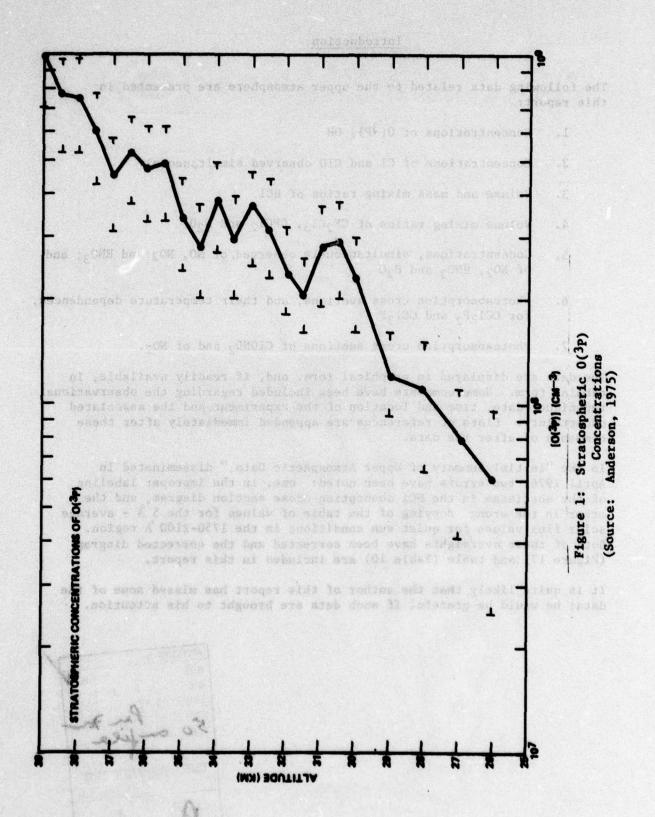


TABLE 1: Stratospheric Concentration of O(3P) (Values read off published graph)

Altitude	O(3P) Con	centration:	Datet
	Observed	Experimental ^a Uncertainty	timalT
<u>ka</u>	cm-3	cm-3 cm-3	10/18/14/14
39	1.0(9)b	7.0(8) - 1.3(9)	Solar Za
38.5		5.2(8) - 9.7(8)	
38.0	7.5(8)	5.2(8) - 9.7(8)	Locacion
37.5	6.1(8)	4.2(8) - 7.7(8)	
. 37.0 to Most of Spieser	4 4/0)	3.1(8) - 5.7(8)	Reperine
36.5	5.2(8)	3.7(8) - 6.6(8)	Ducert
36.0		3.3(8) - 6.2(8)	Comments
35.5	4.9(8)	3.3(8) - 6.2(8)	
25 am integrated of \$250	2 4/01	2.4(8) - 4.4(8)	
34.5 alexanded and the	2 2/2)	2.0(8) - 3.7(8)	
34.0	3.9(8)	2.7(8) - 4.9(8)	
33.5	2.9(8)	2.0(8) - 3.9(8)	
33.0	3.7(8)	2.5(8) - 4.6(8)	
32.5	3.1(8)	2.3(8) - 4.4(8)	
32.0	2.4(8)	1.8(8) - 3.2(8)	
31.5	2.0(8)	1.6(8) - 2.6(8)	
31.0 a die Earth's 0.16		1.9(8) - 3.7(8)	, masteha
30.5	01 A K+125 2.9(8)	2.0(8) - 3.8(8)	tratosph
30	2.3(8)	1.6(8) - 3.0(8)	
29	1.2(8)	9.2(7) - 1.6(8)	
28	1.1(8)	6.4(7) - 1.5(8)	
27	8.1(7)	4.1(7) - 1.1(8)	
26	6.0(7)	2.5(7) - 9.5(7)	

(Source: Anderson, 1975)

Experimenter's uncertainty 1.0(9) refers to 1.0 x 109

Stratospheric Concentrations of O(3P)

Technique: Atomic resonance fluorescence of O(3P) at 1304 Å

Date: 25 November 1974

Time: 10:30 a.m. CST

2.3(8) - 4.6(8) 2.3(8) - 4.6(8) 1.8(8) - 3.2(8)1.6(8) - 2.6(8)

(8)0.E - (8)0.1

Solar Zenith Angle: 560

Location: Palestine, Texas (32° N)

Experimental
Uncertainty: ±30% from 40 km to 29 km increasing to ±60% at 27 km.

Comments: Structure evident in the profile is statistically significant above 30 km.

Profile measured from 40 to 26 km integrated over 500 m intervals above 30 km and over 1 km intervals below 30 km.

Altitude

Anderson, T. G., The Absolute Concentration of O(3P) in the Earth's Stratosphere, Geophys. Res. Lett., 2, 231-234, 1975

Experimenter's uncertainty

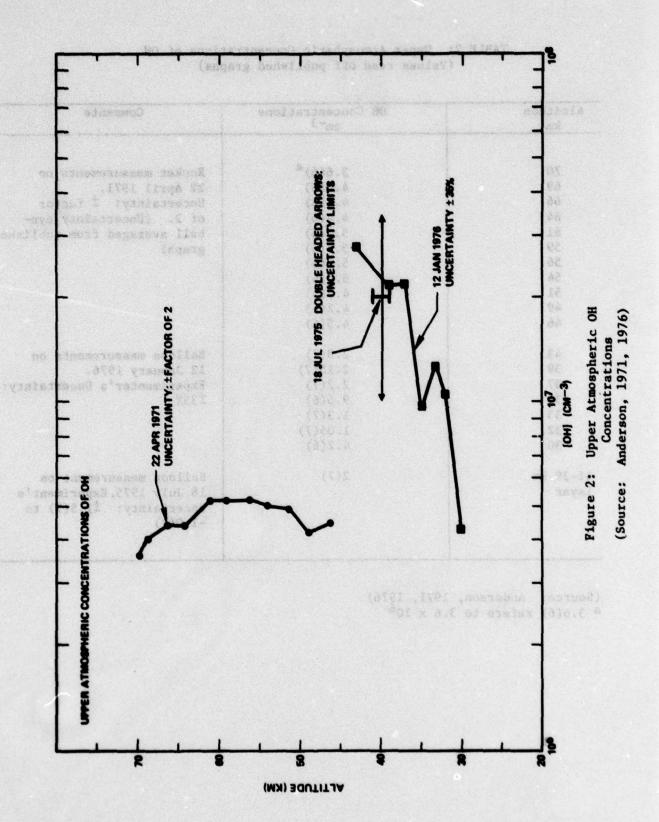


TABLE 2: Upper Atmospheric Concentrations of OH (Values read off published graphs)

Altitude km	OH Concentrations	Comments
70	3.6(6)ª	Rocket measurements on
69	4.0(6)	22 April 1971.
66	4.4(6)	Uncertainty: ± factor
64	4.4(6)	of 2. (Uncertainty eye-
61	5.2(6)	ball averaged from published
59	5.2(6)	graph)
56	5.2(6)	
54	5.0(6)	
51	4.9(6)	
49	4.2(6)	
46	4.5(6)	
43	2.8(7)	Balloon measurements on
39	2.15(7)	12 January 1976.
37	2.2(7)	Experimenter's Uncertainty:
35	9.5(6)	±35x
33	1.3(7)	
32	1.05(7)	
30	4.2(6)	
41-39 km	2(7)	Balloon measurement on
layer		18 July 1975. Experiment's
		Uncertainty: ±1.5(7) to
		-1.0(7)

(Source: Anderson, 1971, 1976) a 3.6(6) refers to 3.6 x 10⁶

Upper Atmospheric Concentrations of OH

Title (importmentar a concertainty)

Resonance fluorescence of OH in the wavelength Technique:

region 3064-3120 Å

22 April 1971

Rocket (Nike-Apache sounding rocket) Platform:

1816 MST Time:

Solar Zenith Angle: 86º 13'

White Sands, New Mexico Location:

Vertical Region

70 to 45 km Sampled:

Uncertainty: Factor of 2 obtained as an eye-ball average from published

graph.

18 July 1975

Platform: Balloon.

Time: Local noon (at launch)

Solar Zenith Angle: 80° (during measurement)

Palestine, Texas (32º N) Location:

Vertical Region

41 to 39 km Sampled:

 $+1.5 \times 10^7$ to -1.0×10^7 cm⁻³ Uncertainty:

(Experimenter's uncertainty)

12 January 1976

Platform: Balloon

Time: Local noon (at launch)

Solar Zenith Angle: 800 (during measurement)

Location:

Palestine, Texas (32º N)

Vertical Region Sampled:

43 to 29 km

testalidad mont elegante lied est to be footbald to be releast.

Uncertainty:

±35% (Experimenter's uncertainty)

Anderson, J. G., Rocket Measurements of OH in the Mesosphere, J. Geophys. Res., 76, 7820-7824, 1971

Anderson, J. G., The Absolute Concentration of OH (X^2 T() in the Earth's Stratosphere, Geophys. Res. Lett., 3, 165-168, 1976

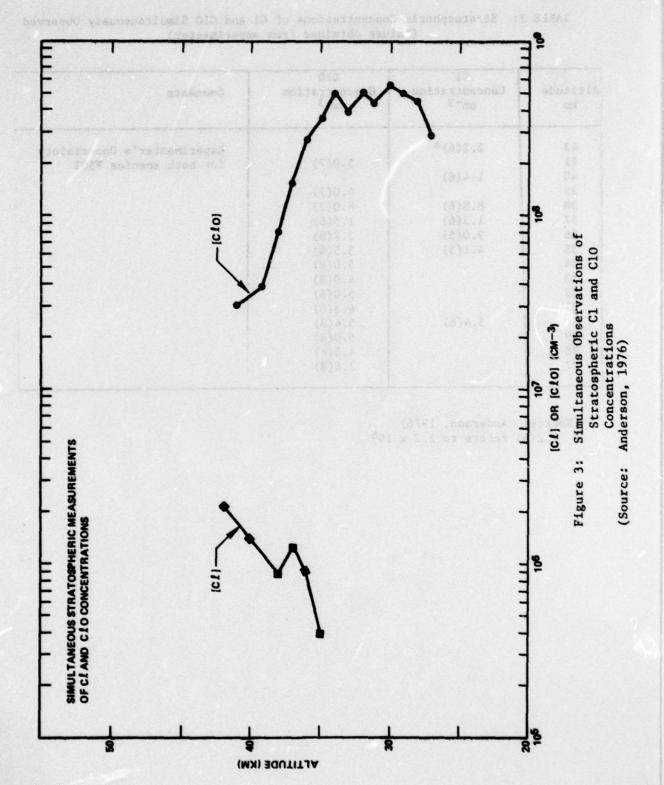


TABLE 3: Stratospheric Concentrations of Cl and ClO Simultaneously Observed (Values obtained from experimenter)

Altitude km	C1 Concentration cm-3	C10 Concentration cm-3	Comments
42	2.2(6)ª		Experimenter's Uncertainty
41		3.0(7)	for both species ±50%
40	1.4(6)		
39		4.0(7)	
38	8.8(6)	8.0(7)	
37	1.3(6)	1.5(8)	
36	9.0(5)	2.7(8)	
35	4.1(5)	3.5(8)	
34		5.0(8)	
33		4.0(8)	
32		5.0(8)	
31		4.3(8)	
30	5.4(8)	5.4(8)	
29		5.0(8)	
28		4.5(8)	
27		2.8(8)	

(Source: Anderson, 1976) a 2.2(6) refers to 2.2 x 106

Simultaneous Stratospheric Measurements of Cl and ClO Concentrations

Technique: Resonance fluorescence (C10 does not fluoresce;

hence it is converted to Cl by adding NO: Cl0 + NO->

C1 + NO2 , and resulting C1 detected at 1188 Å)

Platform: Balloon

Two instruments, one for Cl and the other for ClO,

launched simultaneously on the same balloon

Date: 28 July 1976

Time: 12 Noon CDT

Solar Zenith Angle: 160

Location: Palestine, Texas (32°N)

Uncertainty: ±50% for both species (Experimenter's uncertainty)

Anderson, J. G., A Simultaneous Measurement of Cl and ClO in the Earth's Stratosphere, Paper presented at the International Conference on the Stratosphere and Related Problems, Logan, Utah, September 15-17, 1976.

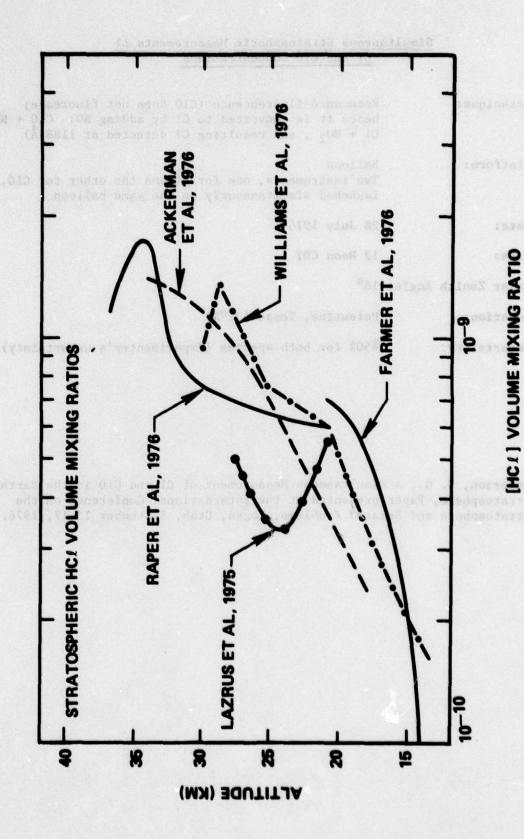


Figure 4: Stratospheric HCl Volume Mixing Ratios

12

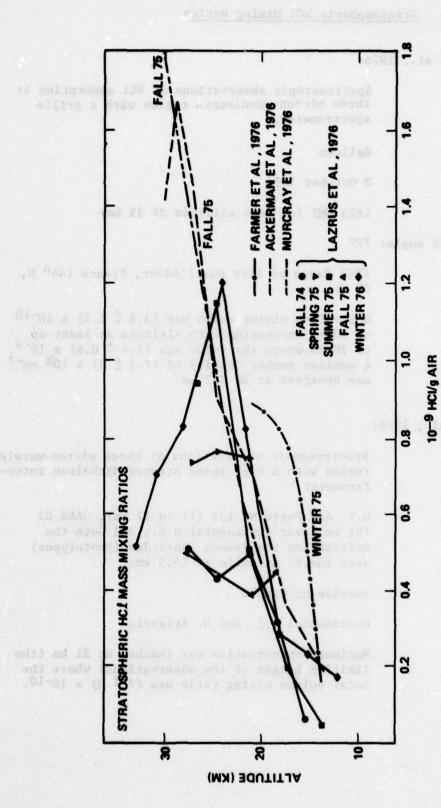


Figure 5: Stratospheric HCl Mass Mixing Ratios

Stratospheric HCl Mixing Ratios

1. Ackerman et al., 1976:

Technique: Spectroscopic observations of HCl absorption at

three micron-wavelength region with a grille

spectrometer.

Platform: Balloon

Date: 2 October 1975

Time: 1623 GMT (at peak altitude of 35 km)

Solar zenith angle: 770

Location: CNES Range at Aire sur l'Adour, France (44° N,

00 W)

Comments: HCl volume mixing ratio was $(3.8 \pm 1.5) \times 10^{-10}$

at 20 km increasing with altitude at least up to 30 km where the value was $(1.4 \pm 0.6) \times 10^{-9}$. A maximum number density of $(7.2 \pm 3) \times 10^{8}$ cm⁻³

was observed at 24 ± 2 km.

2. Farmer et al., 1976:

Technique: Spectroscopic observations at three micron-wavelength

region with a high speed stepped Michelson inter-

ferometer

Platform: U.S. Air Force NC 135 (11 to 12 km), NASA U2

(21 km) over Continental U.S.; and both the British and the French Concordes (prototypes)

over the N. Atlantic at 15.5 km.

Time: Sunrise or Sunset

Location: Continental U.S. and N. Atlantic

Comments: Maximum concentration was reached at 21 km (the

limiting height of the observations) where the

local volume mixing ratio was $(7 \pm 1) \times 10^{-10}$.

Stratospheric HCl Mixing Ratios (cont'd)

Lazrus et al., 1975:

Technique: In situ sampling using filter capture.

Platform: Balloon

Time: Fall 1974, generally in early morning.

Location: Hollomon Air Force Base, New Mexico

The volume mixing ratios were 5.5 x 10^{-10} at 21 km, 3.4 x 10^{-10} at 24 km, 5.1 x 10^{-10} at Comments:

26.4 km and 4.0 x 10-10 at 27.5 km.

Lazrus et al., 1976:

See Lazrus et al., 1975. The times of experiments are given in the

Raper et al., 1976:

See Farmer et al., 1976 Technique:

Balloon A Managara Ma Platform:

Date: September 1975 and May 1976

Palestine, Texas (32° N) Location:

Comments:

The HCl volume mixing ratio increased from about 6×10^{-10} at 20 km to a maximum of about 1.7 \times 10^{-9} at 34-35 km and fell off rapidly thereafter to less than 4×10^{-10} at 40 km.

Williams et al., 1976:

Technique: Spectroscopic observations at three micron-

wavelength region with a grating spectrometer at

float altitude, 30 km.

Platform: Balloon

Date: 16 December 1975

Stratospheric HCl Mixing Ratios (cont'd)

Time: Sunset

Location: Hollomon Air Force Base, New Mexico

Comments; The volume mixing ratio increased from 1.5×10^{-10} to 1.2×10^{-9} in the 13.4 to 27 km altitude range.

Ackerman, M., D. Frimout, A. Girard, M. Gottignies, and C. Muller, Stratospheric HCl From Infrared Spectra, Geophys. Res. Lett., 3, 81-83, 1976.

Farmer, C. B., O. F. Raper, and R. H. Norton, Spectroscopic Detection and Vertical Distribution of HCl in the Troposphere and Stratosphere, Geophys. Res. Lett., 3, 13-16, 1976.

Lazrus, A. L., B. W. Gandrud, R. N. Woodard, and W. A. Sedlacek, Stratospheric Halogen Measurements, Geophys. Res. Lett., 2, 439-441, 1975, as quoted in "The Effect of Fluorocarbons on the Concentration of Atmospheric Ozone" by the Technical Panel on Fluorocarbon Research, Manufacturing Chemists Association, 1825 Connecticut Ave., N.W., Washington, D.C. 20009, 1 March, 1976.

Lazrus, A. L., B. W. Gandrud, R. N. Woodard, and W. A. Sedlacek, Variability of Stratospheric Hydrogen Chloride, Private Communication, 1976.

McCarthy, R., An Industry View of the Scientific Aspect of the Fluorocarbon/ Ozone Issue, Paper presented at the International Conference on the Stratosphere and Related Problems, Logan, Utah, September 15-17, 1976.

Raper, O. F., C. B. Farmer, and R. A. Toth, The Vertical Distribution of HC1 in the 20-40 km Region of the Stratosphere, Paper presented at the International Conference on the Stratosphere and Related Problems, Logan, Utah, September 15-17, 1976.

Williams, W. J., J. J. Kosters, A. Goldman, and D. G. Murcray, Measurement of the Stratospheric Mixing Ratio of HCl using Infrared Absorption Technique, Geophys. Res. Lett., 3, 383-385, 1976.

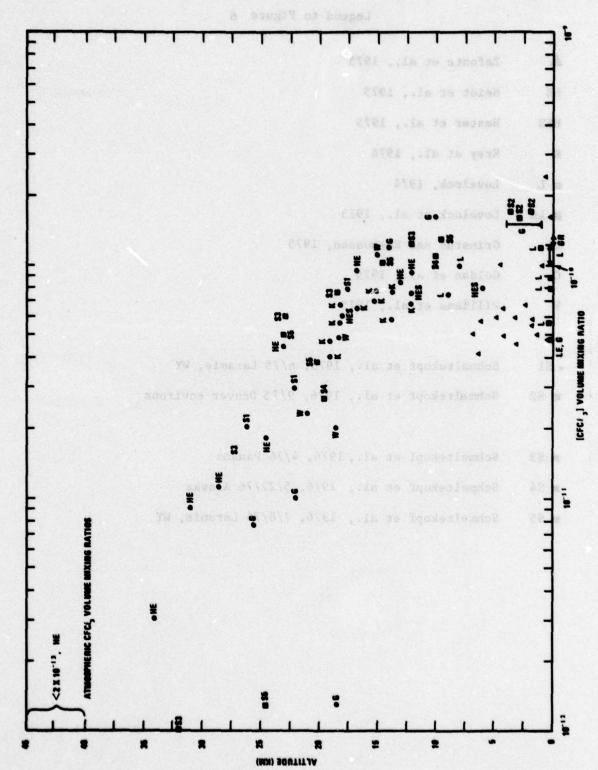


Figure 6: Atmospheric CFCl₃ Volume Mixing Ratios

Legend to Figure 6

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\blacksquare	Zafonte et a	1. 19/5

HE Heidt et al., 1975

HES Hester et al., 1975

K Krey et al., 1976

L Lovelock, 1974

LE Lovelock et al., 1973

GR Grimsrud and Rasmussen, 1975

G Goldan et al., 1975

W Williams et al., 1975

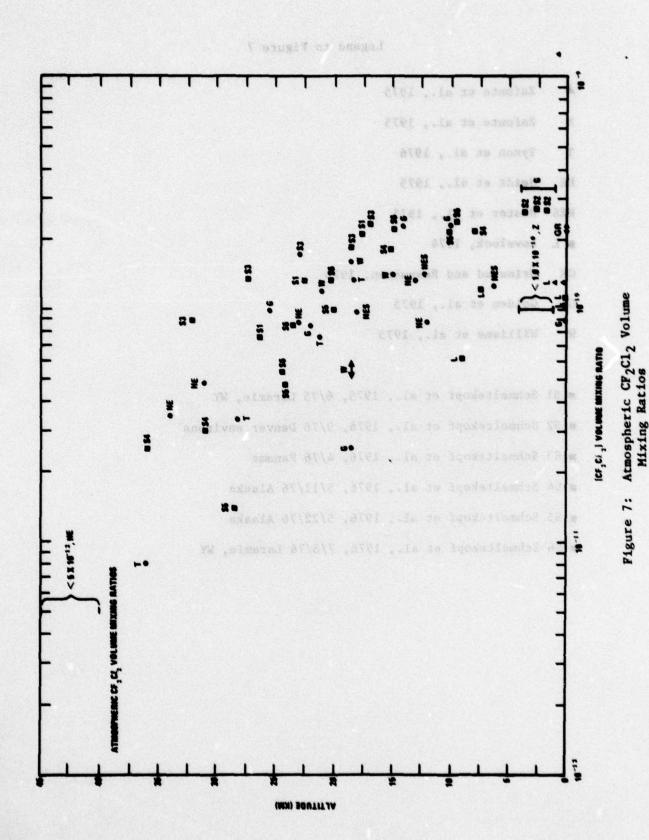
- S1 Schmeltekopf et al., 1975, 6/75 Laramie, WY

S2 Schmeltekopf et al., 1976, 9/75 Denver environs

S3 Schmeltekopf et al., 1976, 4/76 Panama

■ S4 Schmeltekopf et al., 1976, 5/22/76 Alaska

S5 Schmeltekopf et al., 1976, 7/8/76 Laramie, WY



Legend to Figure 7

- A Zafonte et al., 1975
- Z Zafonte et al., 1975
- T Tyson et al., 1976
- HE Heidt et al., 1975
- HES Hester et al., 1975
- L Lovelock, 1974
- GR Grimsrud and Rasmussen, 1975
- G Golden et al., 1975
- W Williams et al., 1975
- m S1 Schmeltekopf et al., 1975, 6/75 Laramie, WY
- S2 Schmeltekopf et al., 1976, 9/76 Denver environs
- S3 Schmeltekopf et al., 1976, 4/76 Panama
- S4 Schmeltekopf et al., 1976, 5/11/76 Alaska
- S5 Schmeltekopf et al., 1976, 5/22/76 Alaska
- m S6 Schmeltekopf et al., 1976, 7/8/76 Laramie, WY

TABLE 4: MEASUREMENTS OF FLUOROCARBON VOLUME MIXING RATIOS UNITS: pptv (10⁻¹²v/v)

Comments		Rocket-integrated air	(Ehhalt et al., 1975)	Balloon-borne cryogenic	sampling; gas chromatography with an electron capture	detector Soughamer and	N. 10023.		part with function of the part and part and part of the part of th		Tropopause: 15km (1973)	16.19km (1975) 32°N)	30 Lat Average value from 2 flights 45 Long.	14 Lat.
Date & Location	15103	5/23/73	· · · · · · · · · · · · · · · · · · ·	5/1/14		5/1/14	6/2/75	6/2/75	9/9/73	6/2/75	9/9/73	9/9/73 Palestine, TX (32°N)	5/23/74 36.15 -39.30 Lat. 106.17-106.45 Long.	5/23/74 33.10 -34.14 Lat. 104.30-105.10 Long.
$(\tilde{cr}_2\tilde{cl}_2)$		6 5	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	35		87	١	1	98	ı	133	78	98 ± 18 (110)	140 (140)
(cret ₃₎	9.08	< 0.2	ESS Mile Mile Mile Mile Mile Mile Mile Mile	3		6	ı	18	57	95	8	76	60 ± 4 (57)	75 (75)
Altitude, km		40-45	Sattance	34.0	10.00	31.0	28.6	24.5	23.0	16.9	13.0	12.0	18.3	12.2 (12)
Author		Reidt et al.,	c/m					21			`	year	Hester et al., 1975	s dellar.

Author	Altitude, km	P-11 (QC11)	F-12 (GP2CL2)	Date & Location	Comments
18.2	4.9	80 ± 3 (82)	125 ± 7 (120)	5/23/74 34.45 -33.50 Lat. 106.20-105.00 Long.	Values in parentheses quoted by Reidt et al., 1975
Krey et al.,	13.7	29		4/74	
	15.2	20		9/8/3/8	16 tare (1812)
	16.8	65	AC.	4190,322	CONTRACTOR
	18.3	57			
	19.2	17			Compressed air sample
	12.2	69		10/74	Gas chromatography analysis
22	13.7	11	1	S_01-40C/	
	15.3	71		SPECIAL SPECIA	Mean concentration averaged
	16.8	99		W(1)(1)	sampling
1	18.3	19			Shirt was entertial and spinor of the state
	19.2	14	35	263.000	Services proper attackings
Lovelock,	Surface	79.8	101.7	6/74, 7/74	Gas chromatography
The state of the state of		2.0.2	* >	W. Ireland	10 0 00 10 10 10 10 10 10 10 10 10 10 10
		99.0	113.2	N. Atlantic	all nowers.
Andena		101-119		6/74 Central England	
	2.000	57	WINE TOOMSHIP	9/74 Capetown, S. Africa	

	Altitude, km	(QC13)	(CP2C12)	Date & Location	Comments
	6	75	09		Tropopause between 7.5 and 9 km
	7.5	100	122	1 1 1 1 1 1	
	1	118	128		the second cards again the decident
Lovelock et al., 1973	Surface	64		11/71, 12/71 500N-60°S	Oceanographic cruise of Shackleton Latitude
	27. 1	0.70	4		Aerial concentration averaged over 50°N-60°S. Concentration ranges from 70 pptv at 50°N to 38 pptv at 60°S.
Schmeltekopf et al., 1975	26.2 ± 1	20	75 ± 5	6/75 Laramie, WY	Balloon-borne stainless steel grab samplers
	22,3 ± 0.7	30+3	135 ± 10	9/15/48	Electron capture detector/gas chromatography
0	17.7 ± 0.5	80 ± 10	210 ± 10	8198132	
	Surface	120-130	210-230	May 1975	Gas chromatography/
and Rasmussen, 1975	52.7	1.6	740 J. 70	6716	Walues as quoted by Sze and
Goldan et al.,	Surface	5 7 87	90 ± 10	8/75	
	1-4	150	330 > 4 3	87.75	Values as quoted by Sze and Wu, 1976.
YMENOL	-10	160 ± 15	230 ± 30	. redidence 4 5//8	\$5 populació
	14	120 ± 15	225 ± 30	8/75	The second secon

18.5 1832 2014 m. sc apr'' 22		(1237)	(Ur 2012)		
22	10	1.3 + 0.7	25 ± 3	8/75	and the ost of Bostoap as south
22		- 0.3		Saskatchewan	
	932	10 ‡ 1	84 ± 8	8/75	
7632	2	7.6 ± 1	100 ± 10	8/75	ANTOR TONGONOUS OL SER BUT
Williams 21	5	23	120	9/26/75	See Williams et al., 1975.
et al., 1975 18.5	2.0 %	67	160	9/26/75	
18.5	•	20	09-05	8/12/68	SERVICE ORDERS SERVICE STATES
	H	110	140	9/26/75	のでは、
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Schmeltekopf 1.75	75	170	270	51/6	Electron capture detector/gas
et al., 19/0 2.5	2	160	270	Denver	
passipace of 3.5	5	170	260	Environs	
-		118			(Values read off the grapits
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Schweltelopf 12 125 230 4/76 Electron capture detector/gas et al., 1976 18.5 77 185 Panama chromatography 23 60 1175 Panama chromatography 175 Panama 27 15 135 Panama 27 135 Panama Pan	a fi com mi a monament sera son a consta co menta despetador.		(carra)	(CF ₂ C1 ₂)	Date and Location	Coments
18.5 77 185 Panama 23 60 175 27 15 135 32 1 90 8 220 5/11/76 115 30 41aska 36 25 272/76 110 220 5/22/76 24 47 41 28.5 14	Management for the rest activates to best finding	12	125	230		
23 60 175 27 15 135 32 1 90 8 220 5/11/76 115 180 Alaska 36 27 100 Alaska 24 47 28.5 114	THE THE STREET, IS SHEET THE	18.5	<i>n</i>	185		
27 15 1 90 8 220 5/11/76 15 180 Alaska 31 30 Alaska 10 110 220 5/22/76 19.5 27 100 Alaska 24 47 47 28.5 14		23	8	175		\
32 1 90 8 220 5/11/76 115 180 Alaska 34 30 25 36 25 19.5 27 100 Alaska 24 47 28.5 114		27	15	135		
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14		19.5	77	100	Alaska	
		24		1.7		
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THE REAL PROPERTY.

Author	Altitude	F-11 (CPC1 ₂)	F-12 (CF ₂ C1 ₂)	Date and Location	Coments
Schmeltekopf	9.5	130	240	9//8//	Electron capture
1 41., 1970	14.5	100	220	Laramie, WY	chromatography
	20.3	39	135		ge vil 1944 -
	23.3	\$	48	3 3 5 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6 6	(Values read off the graphs
	24.5	1.3	\$5	21.5313.6	province by the experimenters)
Tyson et al., 1976	18.3		132 ± 3	3/23/76, Oregon 42º 08'N, 117º 15'W	Cryogenic Sampling by U-2 aircraft at 18.3 and 21.3
26	21.3		84 ± 1	3/23/76, Oregon 42º 13'N, 117º 20'W	km and by balloon at 28.3 and 35.9 km.
)	21.3		73 ± 1	2/20/76, Calif. 37º 45'N, 120º 08'W	
	21.3		73 ± 2	3/11/76, Calif. 36° 54'N, 119° 38'W	
	21.3	6	74 ± 10	5/14/76, Canada 55º 50'N, 67º 45'W	Astrodomatica-pos pata es bushivesq
	21.3		76 ± 2	5/14/76, Canada 61º 00'N, 68º 35'W	educing and the least people?)
	Mean at: 21.3		76 ± 3	authorização	Constraint and the state of the
Mercen	28.3		34 ± 3	1/23/76, Texas 32º 08'N, 92º 26'W	The second secon
	35.9	T H	8 ± 2	1/23/76, Texas 31º 26'N, 94º 05'W	

Zafonte et al., 0.426 160 130 2/23/73, Riverside engined aircraft Whole air samples by twin-engined aircraft 1975 0.610 240 1300 Califf., Rialto Rialto engined aircraft 1.372 150 < 100 Rialto Rialto Rialto Rialto Rialto Aidel 1.372 1.00 Rialto Rialto Rialto 1.00 Rialto 1.00 1.00 Rialto 1.00 1.00 Rialto 1.00 1.0	nte et al.	-		17 9		
0.914 100 130 Califf., Rialto 1.372 150 <100 Rialto 1.829 58 <100 Rialto 2.348 67 <100 Rialto 3.048 59 <100 Rialto 4.267 65 <100 Rialto 1.372 49 <100 3/7/73, 1.372 49 <100 3/7/73, 1.3829 55 <100 3/7/73, 4.268 50 <100 Califf. 6.248 42 <100 Califf. 1.625 6.060		0.426	160	130	2/23/73, Riverside	Whole air samples by twin-
0.914 100 130 Rialto 1.372 150 < 100	2.4	0.610	240	1300	Calif., Rialto	בחלדות פיונופור
1.372 150 < 100		916.0	100	130	Rialto	CFI
1.829 58 < 100		1.372	150	< 100	Rialto	Marka Marka Marka Marka Marka Marka
2.348 67 < 100 Rialto 3.048 59 < 100	462	1.829	58	< 100	Rialto	81 13 16 pt 1 15 pt 1
59 < 100 Rialto 65 < 100 3/7/73, 48 < 100 3/7/73, 49 < 100 8ialto, Calif. 50 < 100 6100 50 < 100 6100 58 < 100 58 < 100 58 < 100 59 < 100 59 < 100 59 < 100		2.348	19	<100	Rialto	143 143 143 143 143 143 143 143 143 143
4.267 65 < 100 Rialto 0.457 48 < 100		3.048	59	< 100	Rialto	THE STATE OF THE S
0.457 48 < 100		4.267	65	<100	Rialto	G A A A A A A A A A A A A A A A A A A A
48 < 100 3/7/73, 49 < 100 Rtaito, Calif. 55 < 100 C 100 C 100 50 < 100 C 100 58 < 100 C 100 51 < 100	46.0				11 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	PALL SANGE
49 < 100		0.457	87	< 100	3/7/73,	Whole air samples by twin-
55		1.372	67	< 100	Rialto, Calif.	per open date transcaperto
46 < 100 be 60 and 90 pptv, respectively. 50 < 100 fively. 58 < 100 fively. 51 < 100 fively.		1.829	55	< 100	n sin h	background levels of F-11
50 < 100 58 < 100 42 < 100 51 < 100		3.048	97	< 100		be 60 and 90 pptv, respect-
58 < 100 42 < 100 51 < 100		4.268	95	× 100		TVELY.
58 < 100 42 < 100 51 < 100		4.877	09	< 100	100 mm (100 mm) (100	4 A (2) A (
51 < 100		960.9	88	× 100	在 1	
21		6.248	42	< 100	dand start start surt surt surt	
		901.9	51	< 100	entition of the second	

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- Ehhalt, D. H., L. E. Heidt, R. A. Lueb, and E. A. Martell, "Concentrations of CH₄, CO, CO₂, H₂, H₂O and N₂O in the Upper Atmosphere," J. Atm. Sci., 32, 163-169, 1975.
- Heidt, L. E., R. Lueb, W. Pollock, and D. H. Ehhalt, "Stratospheric Profiles of CC13F and CC12F2," Geophys. Res. Lett., 2, 445-447, 1975.
- Hester, N. E., E. R. Stephens, and O. C. Taylor, "Fluorocarbon Air Pollutants," III, Environ. Sci. and Technol., 9, 875-876, 1975.
- Krey, P. W., R. J. Lagomarsino, and J. J. Frey, "Stratospheric Concentrations of CC13F in 1974," J. Geophys. Res., 81, 1557-1560, 1976.
- Lovelock, J. E., "Atmospheric Halocarbons and Stratospheric Ozone," Nature, 252, 292-294, 1974.
- Lovelock, J. E., R. J. Maggs, and R. J. Wade, "Halogenated Hydrocarbons in and over the Atlantic," Nature, 241, 194ff, 1973.
- Schmeltekopf, A. L., P. D. Goldan, W. R. Henderson, W. J. Harrop, T. L. Thompson, F. C. Fehsenfeld, H. I. Schiff, P. J. Crutzen, I. S. A. Isaksen, and E. E. Ferguson, "Measurement of Stratospheric CFCl₃, CF₂Cl₂, and N₂O," Geophys. Res. Lett., 2, 393-396, 1975.
- Schmeltekopf, A. L., et al., Private Communication, 1976.
- Sze, N. D., and M. F. Wu, "Measurements of Fluorocarbons 11 and 12 and Model Validation: An Assessment," to appear in Atmospheric Environment, 1976.
- Tyson, B. T., R. B. Brewer, J. A. Arveson, and J. F. Vedder, "Concentrations of Freon 12 and Nitrous Oxide in the Stratosphere," Paper presented at the International Conference on the Stratosphere and Related Problems, Logan, Utah, September 15-17, 1976.
- Williams, W. J., J. J. Kosters, A. Goldman, and D. G. Murcray, "Simultaneous Stratospheric Measurements," Paper presented at the fall AGU Meeting, San Francisco, December 1975.
- Zafonte, L., N. E. Hester, E. R. Stephens, and O. C. Taylor, "Background and Vertical Atmospheric Measurements of Fluorocarbon-11 and Fluorocarbon-12 over Southern California," Atm. Environ., 9, 1007-1009, 1975.



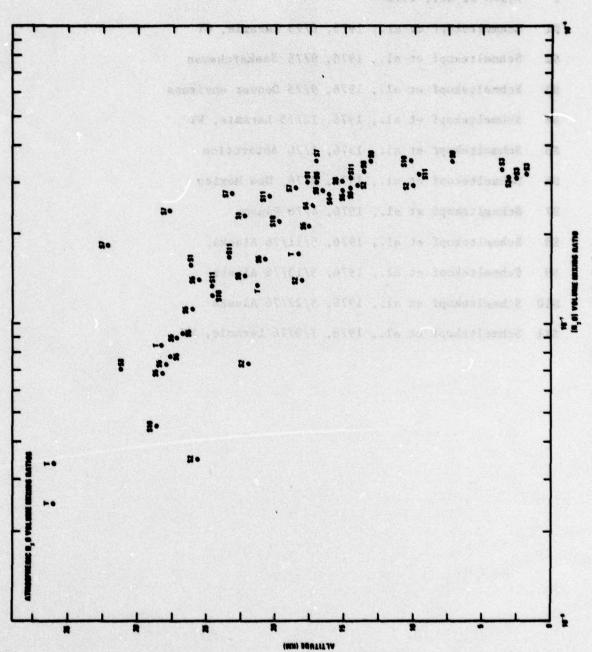


Figure 8: Atmospheric A20 Volume Mixing Ratios

Legend to Figure 8

- T Tyson et al., 1976
- Sl Schmeltekopf et al., 1975, 6/75 Laramie, WY
- S2 Schmeltekopf et al., 1976, 8/75 Saskatchewan
- S3 Schmeltekopf et al., 1976, 9/75 Denver environs
- S4 Schmeltekopf et al., 1976, 12/75 Laramie, WY
- S5 Schmeltekopf et al., 1976, 1/76 Antarctica
- S6 Schmeltekopf et al., 1976, 2/76, New Mexico
- S7 Schmeltekopf et al., 1976, 4/76 Panama
- S8 Schmeltekopf et al., 1976, 5/11/76 Alaska
- S9 Schmeltekopf et al., 1976, 5/13/76 Alaska
- S10 Schmeltekopf et al., 1976, 5/22/76 Alaska
- S11 Schmeltekopf et al., 1976, 7/8/76 Laramie, WY

TABLE 5: ATMOSPHERIC N20 VOLUME MIXING RATIOS Units: ppbv (10-9 v/v)

Author	Altitude km	(M ₂ 0) ppbv	Date & Location	Comments	
Schmeltekopf et al., 1975	17.7 ± 0.5 22.3 ± 0.7 26.2 ± 0.1	300 ± 20 230 ± 20 160 ± 20	6/75 Laramie, WY	Bailoon-borne grab sampling; electron capture detector/gas chromatography. Values published by experimenters	electron graphy. iters
Schmeltekopf et al., 1976	10 14 18 22 25.5	290 290 140 35	8/75 Saskatchevan		
	1.75 2.5 3 3.5 1.6 17.25	320 330 250 250 250 250	9/75 Denver Environs 12/75 Laramie, WY		
	15 17.5 20.5 20.5 25.5 26.5 26.5 27.5 27.75	280 215 1165 113 23 23 23 23	1/76 Antarctica		
Equiva e Consider action in the de-	17	300	2/76 New Mexico	Control of pages and pages and pages of the	national otherwises or Processes

Author	Altitude	(M ₂ 0) ppbv	Date & Location	Comments	-:
Schmeltekopf et al., 1976	11 18.5 23.5 27.5	340 230 238 185	4/76 Panama	Balloon-borne grab sampling; electron capture detector/gas chromatography. Values published by experimenters	
•	7.5 115 31 36	356 300 20 25	5/11/76 Alaska		
town to Miller to the Miller	13.5 14.5 16.5	33.00 33.00 33.00 33.00 33.00 33.00	5/13/76 Alaska		
32	10.5 24.5 28.5	380 220 45 55 55	5/22/76 Alaska		
	9.5 14.5 20.25 23.25 24.50	320 310 265 168 135	7/8/76 Laramie, WY		
Tyson et al., 1976	18.3	171 ± 12	3/23/76, Oregon 42º 08'N, 117º 15'W		
Pendantinenta 2147 - Lin 70	21.3	122 ± 7	3/23/76, Oregon 42º 13'N, 117º 20'W	28.3 and 35.9 km. Values provided by experimenters	
100	21.3	117 ± 4	2/20/76, Calif. 37° 45'N, 120° 08'W		
		A SE OF SE SE	Ships of Property and the		

13 13 3 11 1,05	thomas (1)	à ·		GI THÝ			
Comments	Cryogenic Sampling by U-2 aircraft at 18.3 and 21.3 km and by balloon	Values provi					
Date & Location	3/11/76, Calif. 36º 54'N, 119º 38'W	5/14/76, Canada 54° 50'N, 67° 45'W	5/14/76, Canada 61º 00'N, 68º 35'W		1/23/76, Texas 32º 08'N, 92º 26'W	1/23/76, Texas 31º 26'N, 94º 05'W	
(N ₂ 0) ppbv	129 ± 5	143 ± 11	159 ± 9	134 ± 7	85 ± 1	34 ± 2	
Altitude km	21.3	21.3	21.3	Mean at 21.3	28.3	35.9	
Author	Tyson et al., 1976					33	

.

Schmeltekopf, A. L., P. D. Goldan, W. R. Henderson, W. J. Harrop, T. L. Thompson, F. C. Fehsenfeld, H. I. Schiff, P. J. Crutzen, I. S. S. Isaksen, and E. E. Ferguson, "Measurement of Stratospheric CFCl₃, CF₂Cl₂ and N₂O", Geophys. Res. Lett., 2, 393-396, 1975.

Schmeltekopf, A. L., et al., Private Communication, 1976.

Tyson, B. T., R. B. Brewer, J. A. Arveson, and J. F. Vedder, "Concentrations of Freon 12 and Nitrous Oxide in the Stratosphere," Paper presented at the International Conference on the Stratosphere and Related Problems, Logan, Utah, September 15-17, 1976.

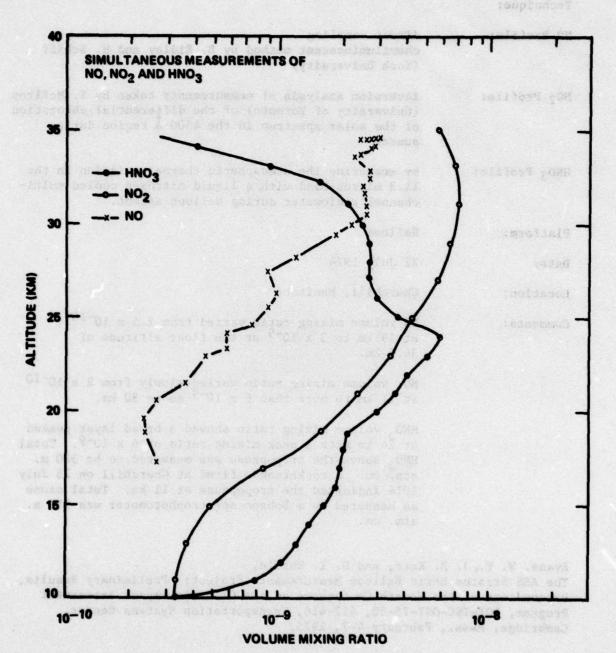


Figure 9: Simultaneous Measurements of NO, NO₂ and HNO₃ in the Stratosphere (Source: Evans et al., 1975)

Simultaneous Observations of NO, NO2 and HNO3

Technique:

NO Profile: direct sampling

chemiluminescent method by B. Ridley and H. Schiff

(York University)

NO2 Profile: inversion analysis of measurements taken by T. McElroy

(University of Toronto) of the differential absorption of the solar spectrum in the 4500 Å region during

sunset

HNO3 Profile: by measuring the atmospheric thermal emission in the

11.3 micron band with a liquid nitrogen cooled multi-

channel radiometer during balloon ascent.

Platform: Balloon

Date: 22 July 1974

Location: Churchill, Manitoba

Comments: NO volume mixing ratio varied from 2.5×10^{-10} at 19 km to 3×10^{-9} at the float altitude of

34.5 km.

NO, volume mixing ratio varied slowly from 2 x 10-10

at 11 km to more than 6 x 10⁻⁹ above 30 km.

 HNO_3 volume mixing ratio showed a broad layer peaked at 24 km with a peak mixing ratio of 6 x 10^{-9} . Total HNO_3 above the tropopause was measured to be 320 m. atm. cm. A rocketsonde fired at Churchill on 23 July 1974 indicated the tropopause at 11 km. Total ozone as measured by a Dobson spectrophotometer was 350 m.

atm. cm.

Evans, W. F., J. B. Kerr, and D. I. Wardle, The AES Stratospheric Balloon Measurements Project: Preliminary Results, Proceedings of the Fourth Conference on the Climatic Impact Assessment Program, DOT-TSC-OST-75-38, 412-416, Transportation Systems Center, Cambridge, Mass., February 4-7, 1975.

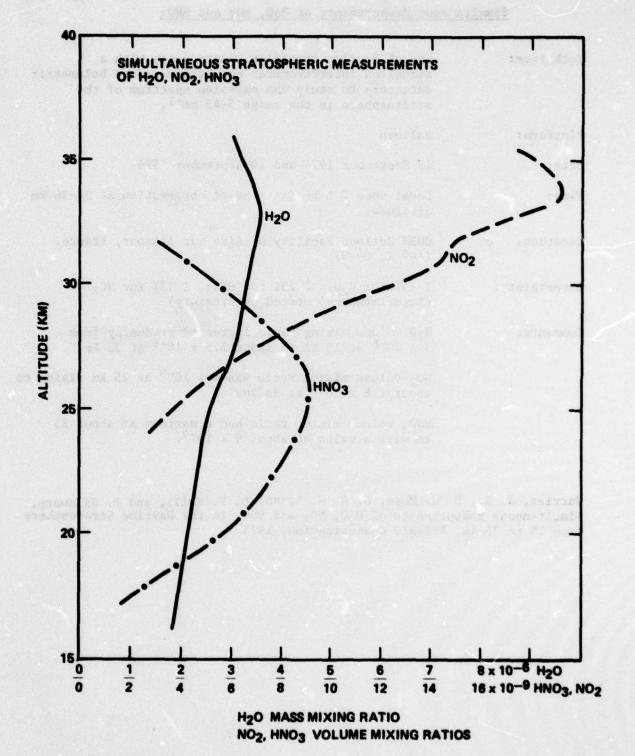


Figure 10: Simultaneous Measurements of NO2, HNO3 and H2O in

the Stratosphere

(Source: Harries et al., 1975)

Simultaneous Measurements of H2O, NO2 and HNO3

Technique: Use of Fourier transform spectroscopy with a

Michelson interferometer and helium-cooled bolometric

detectors to study the emission spectrum of the

stratosphere in the range 5-45 cm-1.

Platform: Balloon.

12 September 1974 and 20 September 1974 Date:

Time: Local noon ± 1 hr (at time of observation at 34-36 km

altitude)

Location: CNES Balloon Facility at Aire sur l'Adour, France.

(440 N, 00 W)

± 15% for H₂O, ± 25% for HNO₃, ± 35% for NO₂ (Experimenters' stated uncertainty) Uncertainty:

H20 volume mixing ratio increased gradually from Comments:

 3×10^{-6} at 15 km to about 5.5 x 10^{-6} at 33 km.

NO₂ volume mixing ratio was 4 x 10-9 at 25 km rising to

about 1.8 x 10-8 at 33 km.

 HNO_3 volume mixing ratio had a maximum at about 25 km with a value of about 9 x 10^{-9} .

Harries, J. E., D. G. Moss, N. R. W. Swann, G. F. Neill, and P. Gildwarg, Simultaneous Measurements of H₂O, NO₂ and HNO₃ in the Daytime Stratosphere from 15 to 35 km, Private Communication, 1975

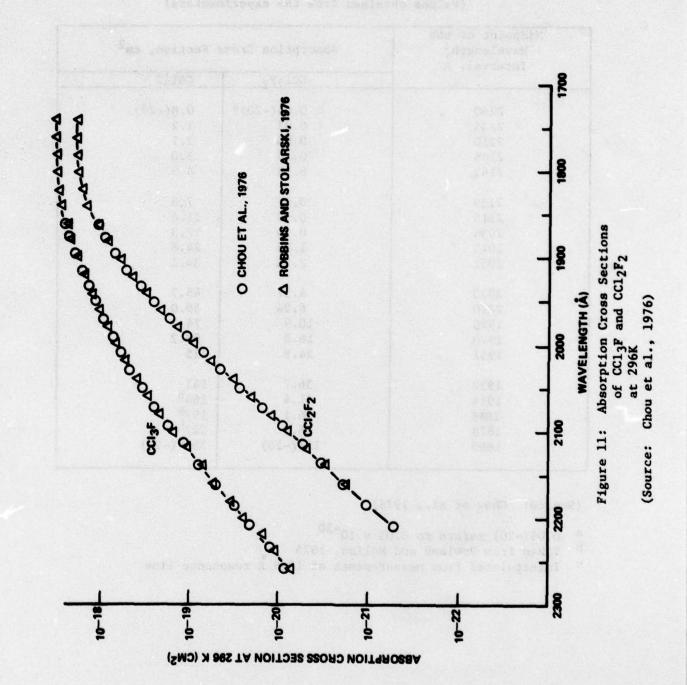


TABLE 6: Absorption Cross Sections for CCl2F2 and CCl3F in the Wavelength Range 1850-2272A at 296K (Values obtained from the experimenters)

Midpoint of the Wavelength Interval, A		Absorption Cross Section, cm ²		
8		CCl ₂ F ₂	CCl ₃ F	
	2260	< 0.05(-20)a	0.8(-20)	
	2235	< 0.05	1.2	
	2210	0.05	2.1	
	2186	0.10	3.0	
8-	2162	0.19	4.9	
	2139	0.32	7.8	
	2116	0.53	11 6	
	2094	0.90	17.3	
	2073	1.53	24.8	
	2051	2.66	34.1	
	2030	4.37	45.7	
	2010	6.96	59.0	
	1990	10.9	74.3	
	1970	16.8	93.2	
	1951	24.9	115	
	1932	36.7	141	
	1914	51.4	164b	
	1896	66.1	197b	
	1878	86.5	227 ^b	
	1860	105(-20)	255 ^c (-20)	

(Source: Chou et al., 1976)

^{0.05(-20)} refers to 0.05×10^{-20}

b Taken from Rowland and Molina, 1975
C Interpolated from measurements at 1849 A resonance line

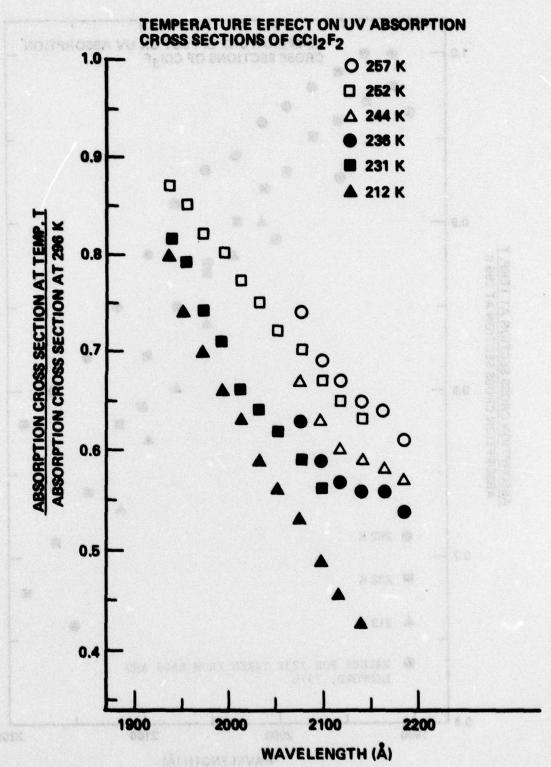
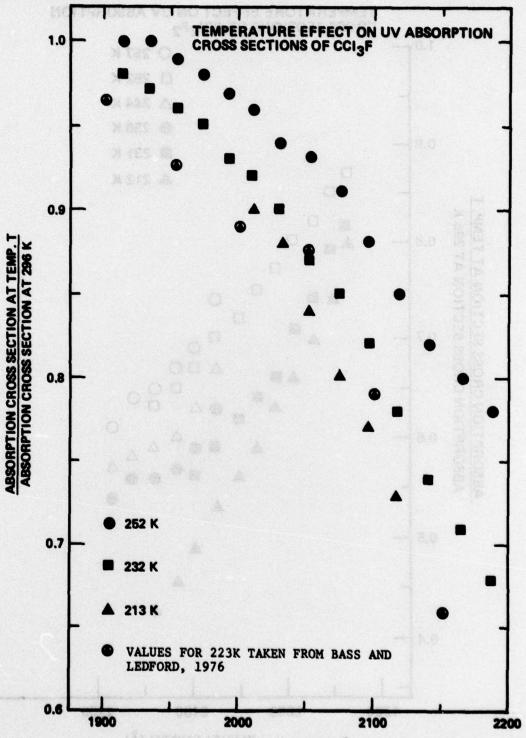


Figure 12: Ratio of the Absorption Cross Section of CCl₂F₂ at Temperature T to that at 296K. (T= 257, 252, 244, 236, 231, 212K) (Source: Chou et al., 1976)



WAVELENGTH (Å)

Figure 13: Ratio of the Absorption Cross Section of CCl₃F at Temperature T to that at 296K. (T=252, 232, 213K) (Source: Chou et al., 1976)

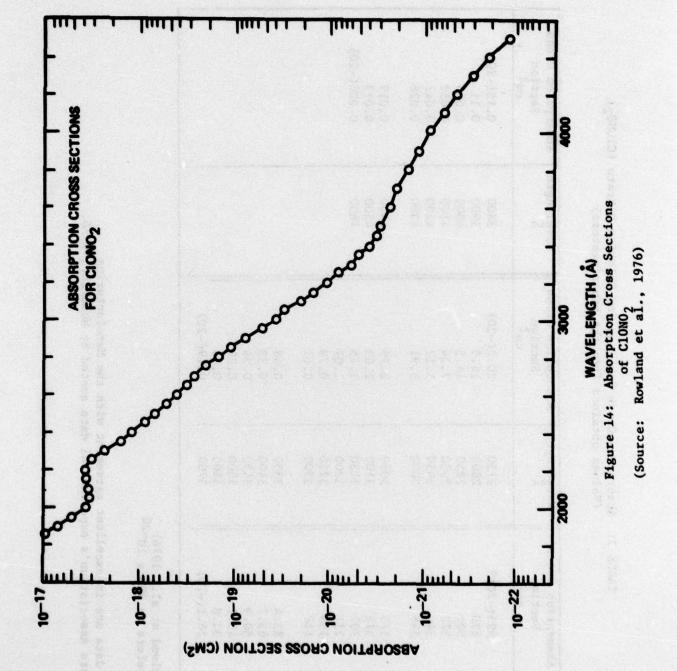


TABLE 7: Absorption Cross Sections for Chlorine Nitrate (ClONO₂) (Values obtained from the experimenters)

wavelength A	Absorption Cross Section cm ²	Wavelength A	Absorption Cross Section cm ²	Wavelength A	Absorption Cross Section cm ²
1860	995(-20)a	2750	20.2(-20)	3800	0.15(-20)
1900	069	2800	14.5	3900	0.11
1950	502	2850	14.5	4000	0.085
2000	372	2900	7.34	4100	0.059
2050	344	2950	5.12	4200	0.042
2100	348	3000	3.91	4300	0.028
2150	375	3050	2.79	0044	0.019
2200	376	3100	2.03	4500	0.013
2250	307	3150	1.45	4600	0.008(-20)
2300	231	3200	1.07		
2350	159	3250	0.79		
2400	2 118	3300	0.61		
2450	85.4	3350	0.48		
2500	65.7	3400	0.38		
2550	50.9	3450	0.34		
2600	40.7	3500	0.29		
2650	32.8	3600	0.23		
2700	26.1(-20)	3700	0.19(-20)		

(Source: Rowland et al., 1976) a 995(-20) refers to 995 x 10-20 These data are in excellent agreement with the Manufacturing Chemists Association's experimental data quoted by McCarthy, 1976. Note:

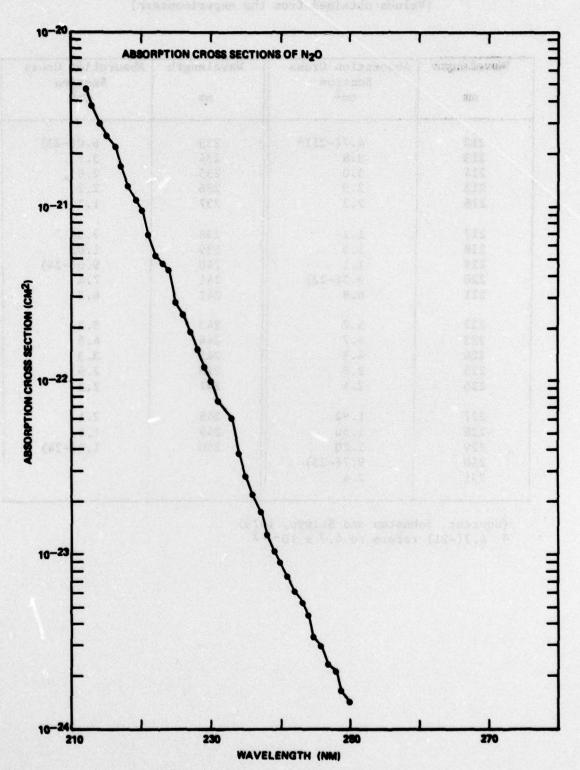


Figure 15: Absorption Cross Sections of N₂0 (Source: Johnston and Selwyn, 1975)

TABLE 8: Absorption Cross Sections of N₂O (Values obtained from the experimenters)

Wavelength	Absorption Cross Section cm ²	Wavelength	Absorption Cross Section cm ²
	CIII		Cu-
212	4.7(-21)ª	233	6.0(-23)
213	3.8	234	3.7
214	3.0	235	2.8
215	2.5	236	2.2
216	2.2	237	1.74 (\$ 0)
217	1.7	238	1.36
218	1.3	239	1.08
219	1.1	240	9.0(-24)
220	9.5(-22)	241	7.4
221	6.9	242	6.2
222	5.2	243	5.2
223	4.7	244	4.5
224	4.3	245	3.3
225	2.8	246	2.9
226	2.4	247	2.3
227	1.92	248	2.1
228	1.50	249	1.6
229	1.20	250	1.4(-24)
230	9.7(-23)		
231	7.4		

(Source: Johnston and Selwyn, 1975) a 4.7(-21) refers to 4.7 x 10-21

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Bass, A. M., and A. E. Ledford, Jr., Ultraviolet Photoabsorption Cross Sections of CF₂Cl₂ and CFCl₃ as a Function of Temperature, Paper presented at the 12th Informal Conference on Photochemistry, National Bureau of Standards, Gaithersburg, MD, June 28-July 1, 1976, as quoted in Chou et. al., 1976.

Chou, C. C., W. S. Smith, H. Vera Ruiz, K. Moe, G. Crescentini, M. J. Molina and F. S. Rowland, The Temperature Dependences of the Ultraviolet Absorption Cross Sections of CCl_2F_2 and CCl_3F , and Their Stratospheric Significance, Private Communication, 1976.

McCarthy, R., An Industry View of the Scientific Aspect of the Fluorocarbon/ Ozone Issue, Paper presented at the International Conference of the Stratosphere and Related Problems, Logan, Utah, September 15-17, 1976.

Robbins, D., and R. Stolarski, as quoted by Chou et. al., 1976.

Rowland, F. S., and M. J. Molina, Chlorofluoromethanes in the Environment, Rev. of Geophys. Sp. Phys., 13, 1-36, 1975.

Rowland, F. S., J. E. Spencer, and M. J. Molina, Stratospheric Formation and Photolysis of Chlorine Nitrate, ClONO₂, Private Communication, 1976.

Johnston, H. S. and G. S. Selwyn, New Cross Sections for the Absorption of Near Ultraviolet Radiation by Nitrous Oxide (N20), Geophys. Res. Lett., 2, 549-551, 1975.

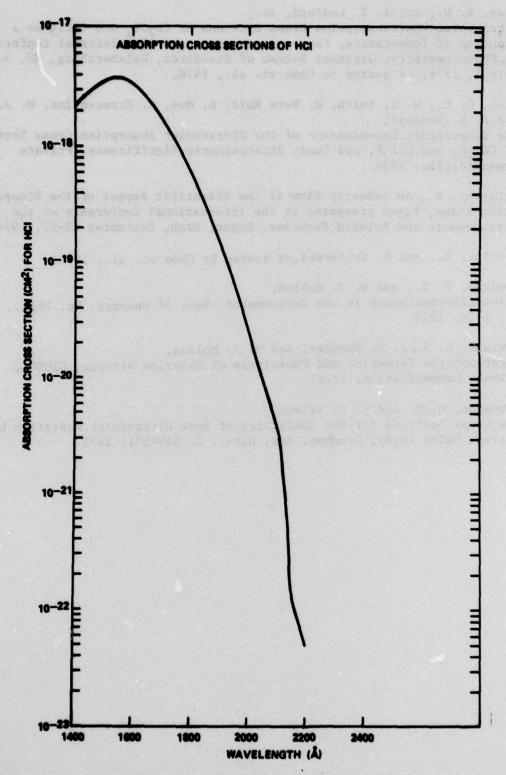


Figure 16: Absorption Cross Sections of HC1 (Source: Inn, 1975)

TABLE 9: ABSORPTION COEFFICIENTS OF HC1 IN THE CONTINUUM 1400-2200 Å

(Source: Inn, 1975)

Xi	(cm ⁻¹ atm ⁻¹)	(x10 ¹⁸ cm ²)	λ (Å)	(cm-1 atm-1)	(x10 ¹⁸ cm ²
1400	56.8	2.11	1775	21.7	0.808
1425	67.4	2.51	1800	15.8	0.588
1450	75.6	2.81	1825	11.6	0.432
1475	87.0	3.24	1850	8.41	0.313
1500	92.7	3.45	1875	5.79	0.215
1525	100.0	3.72	1900	3.90	0.145
1550	102.5	3.82	1950	1.66	0.0618
1575	93.3	3.47	2000	0.688	0.0256
1600	89.1	3.32	2050	0.264	0.00983
1625	79.9	2.97	2100	0.106	0.00395
1650	66.7	2.48	2150	0.0369	0.000137
1675	54.9	2.04	2200	0.0129	0.0000480
1700	43.7	1.63			
1725	35.1	1.31			
1750	29.3	1.09			

 \propto , the absorption coefficient, is defined by I/I_o = exp. $[-(p_0^T)]^T$

where I and I_0 are the transmitted and incident intensity, respectively, p the pressure, T the temperature in K, p_0 = 1 atmosphere, T_0 = 273.15K and 1 is the absorption path length.

o, the absorption cross section, is defined by

5 - X/No

where $N_0 = 2.687 \times 10^{19} \text{ cm}^{-3}$ is the Loschmidt's number.

A is the wavelength.

Inn, E. C. Y., Absorption Coefficients for HCl in the Region 1400 to 2200 Å, J. Atm. Sci., 32, 2375-2377, 1975.

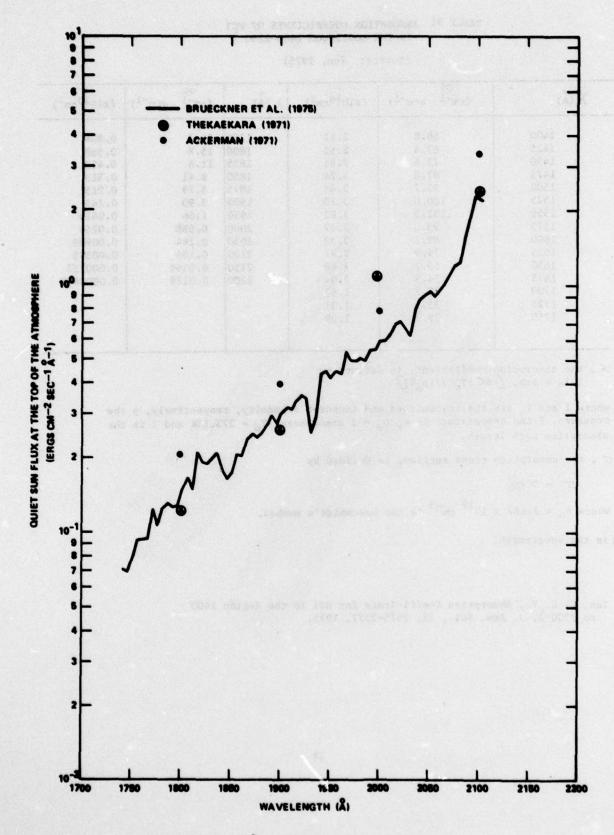


Figure 17: 5 % - Average Solar Fluxes in the 1750-2100 % Range

TABLE 10: 5 Å - AVERAGE SOLAR FLUXES FOR QUIET SUN CONDITIONS (Values provided by Brueckner et al., 1975)

Mid-point of Wave- length interval, (A)	(ergs cm ⁻² s ⁻¹ Å ⁻¹)	Mid-point of Wave- length interval, (A)	FLUX (ergs cm ⁻² s ⁻¹ Å ⁻¹)	
		1927.5	3.49(-1)	
1742.5	7.05(-2)a	32.5	2.50	
47.5	6.89	37.5	2.88	
		42.5	4.34	
1752.5	8.01	47.5	4.52	
57.5	9.18			
62.5	9.28	1952.5	4.18	
67.5	9.40	57.5	4.46	
72.5	1.23(-1)	62.5	4.36	
		67.5	5.33	
1777.5	1.06	72.5	4.93	
82.5	1.25	12.3	4.73	
87.5	1.31	1977.5	4.90	
92.5	1.26		4.70	
97.5	1.27	82.5		
		87.5	5.90	
1802.5	1.46	92.5	5.44	
07.5	1.66	97.5	5.47	
12.5	1.47			
17.5	2.08	2002.5	5.89	
22.5	1.88	07.5	3.34	
22.3	1.00	12.5	6.36	
1827.5	1.83	17.5	6.96	
32.5	1.96	22.5	7.13	
37.5	2.03	TO THE CASE OF MARKET		
	1.75	2027.5	6.69	
42.5	1.62	32.5	6.23	
47.5	1.02	37.5	8.13	
1050 5	,	42.5	8.77	
1852.5	1.74	47.5	9.16	
57.5	2.03			
62.5	2.02	2052.5	9.40	
67.5	2.37	57.5	8.96	
72.5	2.51	62.5	9.50	
		67.5	1.01(0)	
1877.5	2.40	72.5	1.10	
82.5	2.51	A STATE OF THE PARTY OF THE PAR		
87.5	2.69	2077.5	1.19	
92.5	2.93	82.5	1.23	
97.5	2.70	87.5	1.55	
Control Name		92.5	1.89	
1902.5	2.97	97.5	2.27	
07.5	3.17		STATES CONTRACTOR	
12.5	3.10	2102.5	2.25(0)	
17.5	3.39	Burney Charles Co.	The last manufacture where	
22.5	3.59			

(Source: Brueckner et al., 1975) a 7.05(-2) refers to 7.05 x 10⁻²

5 A - Average Solar Fluxes in the 1750-2100 A Range

Date of

Experiment:

September 4, 1973

Agency:

Naval Research Laboratory

Platform:

Rocket (Black Brant VC Rocket)

Instrument:

Double Dispersion Spectrograph

Calibration:

Preflight, ground calibration against a secondary standard deuterium lamp, (continuous emission for >>1680 Å) which was calibrated against NBS absolute standard, a high-power hydrogen arc.

Spectral

Resolution:

0.07 Å

Accuracy:

R. M. S. total error ±20% (down from a factor of 2 or 3

over past measurements).

Method:

Intensity measurements over selected, inactive areas of

the solar disk.

Comments:

A few representative values of the solar fluxes reported by Ackerman (1971) and Thekaekara (1971) have been included in the figure for comparison purposes.

Ackerman, M., Ultraviolet Solar Radiation Related to Mesospheric Processes, in Mesopheric Models and Related Experiments, G. Fiocco (Ed.), 149-159, D. Reidel, Dordrecht, Holland, 1971.

Brueckner, G. E., J.-D. F. Bartoe, O. K. Moe, and M. E. Van Hoosier, Absolute Solar Intensities 1750 Å - 2100 Ä and Their Variations With Solar Activity, E. O. Hulbert Center for Space Research, Naval Research Laboratory, Washington, D.C. 20375, 1975.

Thekaekara, M. P., Solar Electromagnetic Radiation, NASA SP-8005, Goddard Space Flight Center, Greenbelt, Maryland, 1971.